

Thermal Physics, Autumn 2016 CMI

Problem set 5

Due by the beginning of lecture on Monday, Oct 3, 2016

Maxwell relations, Energy equation

1. ⟨7⟩ We used the exactness of $\delta Q/T$ to derive the Maxwell relation

$$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V. \quad (1)$$

Now use the triple product identity and rules for partial differentiation to convert this into a different Maxwell relation

$$\left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p. \quad (2)$$

Hint: On the LHS of (1) regard T as a function of p and S . And use an appropriate triple product identity on the RHS.

2. ⟨7⟩ We derived the (T, V) form of the energy equation in the class. Here derive the (p, T) form. Then specialize to a gas satisfying the ideal EOS $pV = nRT$ and comment on what the energy equation implies.